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Check it out at 3 BC SCIENCE PHYSICS 12 Authors Lionel Sandner Edvantage Interactive Dr. Gordon Gore BIG Little Science Centre (Kamloops) Electrostatics and Electrical Circuits Sample Sample Chapters Project Material, No Final Form Vice-President Marketing: Don Franklin Director Publishing: Yvonne Van Ruskenveld Design/Illustration/Production: Donna Lindenberg bcsienceinteractions.com 4 5 BC Science Physics 12 1 Ve Contents equilibrium Scalars and Vectors In Two Dimensions Statics: Forces Equilibrium Kinematics Uniform Acceleration Shot Motion Pulse and Energy Dynamics Pulse and Pulse Momentum In Two-Dimensional Situations Energy Law Preservation Mechanical Energy Circular Motion and Gravity Movement Circle Gravity and Kepler's Solar System Newton's Law Universal Gravity Electrostatic Static Static Power Electric Field Electric Power Power Energy, Electric Potential, and electrical potential difference Electric Field and Voltage Uniform Field Electric Circuits Current Events History Ohm's Law Kirchoff's Laws Magnetic Forces Basic Ideas on Magnets Magnetic Field Strength , B Magnetic fields and electron electromagnetic induction induced EMF magnetic flow and Faraday's Law induction Answers to a pair of numbered and short answer questions Glossary Index Edvantage Interactive 2013 ISBN Introduction iii 6 7 5 Electrostatic means You must be able to do the following steps by the end of this chapter: Apply coulomb law, to analyze electrical forces Analyze electrical fields and their impact on charged objects Calculate electrical potential energy and changes in electrical energy Apply electrical potential concepts to analyze situations, related to point charges, apply the principles of electrostatic states in different situations Until the end of this chapter , you should know what the following key terms mean: attract the number of theories of the diuturni wires Coulomb's law electric field electric field force electric power electric potential power difference electric potential energy electrons electrostatic electrostatic electrostatic elements simple charge induction insulators law retention fee to want off static electrical voltage To the end of the section, you should be able to and know when to use the following formulae: $Q_1 Q_2 F = k \frac{1}{r^2} \frac{2}{2} E F Q = E = \Delta E Q r = \Delta E d Q Q k r 1 2 p = P = Q = k$ In this chapter you will investigate electrostatic principles such as electrical field interactions modelled in this figure. Edvantage Interactive 2013 ISBN Chapter 5 Electrostatic PROJECT 261 8 5.1 Static Electric Charges Warm Up Place meter stick on watch glass. Rub the inflated balloon on the hair and bring it close to a meter stick. Note the result. Describe and identify the reason for what you notice. Attraction and deterrence forces If your hair is dry and you comb it fast, your comb will attract not only your hair, but also bits of dust, paper or thread. The comb is probably made of plastic, but different types of material will produce the same effect. Already 600 m.c., the Greeks observed the force of amber when it was rubbed with a cloth. Amber is fossilized resin from trees that Greeks use for decoration and trading. Of course, magnets are also leveraging power, but they only attract some metal elements such as iron, nickel and cobalt, and some of their alloys. Amber, if rubbed with a cloth, will attract small pieces just about something. In the late 1500s, Englishman Dr. William Gilbert was curious about this interesting property of amber, and he did many experiments with it and other materials. Gilbert discovered that many materials, if rubbed with some fabrics, could be electrified. Words like electrified, electricity, electrons, and electronics come from the Greek word amber, which was electron. In the early 1700s, Charles du Fay, a French scientist, was probably the first person to figure out that there were two types of electricity. He pointed out that if two glass rods were rubbed with silk and brought to each other side by side, they would repel each other. Repel means to push back. Two amber bars rubbed with fur also repel each other. However, if an electrified amber bar was brought closer to an electrified glass rod, the two bars could tie each other. Du Fay correctly concludes that there must be two types of electricity. Later in the 1700s, Benjamin Franklin named this positive electricity and negative electricity. By convention, a glass rod rubbed with silk is said to be a positive charge. The amber rod rubbed with wool or fur has a negative charge. In classroom experiments, a good way to get a positive fee is to rub the acetate plastic strip with cotton. A negative charge can be easily obtained by rubbing a strip of vinyl plastic with wool or fur (Figure 5.1.1). Chapter 262 5 Electrostatic project Edvantage Interactive 2013 ISBN 9 Image Two charged acetate plastic strips (+), loosely hanging from the support rod, repel each other. Two charged strips of vinyl (-) also repel each other. However, a pay acetate strip will attract a pay vinyl strip. Since electrical charges for electrified items do not move, they are called static charges or Electricity. Static shall be stationary or stationary. Charged object will attract any neutral body. A neutral body is one for free. It will also attract the otherwise charged body, but it will repel another body to do the same charge. Authorities with the same charge to repel each other. Bodies with opposing accusations attract each other. A neutral body attracts either a positively charged body or a negatively charged body. Elementary Atomic Structure Image of a simple planetary model of atoms John Dalton's famous atomic theory assumes that the whole issue consists of inseparable particles. A very important experiment by Ernest Rutherford showed that the atom actually had some internal structure on it. He could show that the atom had a core that concentrated a positive charge. Since the atom as a whole is neutral, it must be negatively charged the substance should somehow be distributed through the nucleus. Negatively charged particles were first identified by English physicist JJ Thomson. They were later called electrons. A simplistic view of the atom, as pictured in rutherford planetary model shows the core of the atom with its positive charge, surrounded by negatively charged electrons. Positively charged particles in the nucleus are protons. The picture also shows neutrons, but they were not discovered until an English physicist named James Chadwick, a modern-day Rutherford, added this particle to the list of subatomic particles. Neutrons carry no electrical charge, and their mass is only slightly higher than the proton. Electrons are much less massive than protons or neutrons. The mass of the protons is kg, which is 1836 times the mass of the electron. The smallest atom is a hydrogen atom. This is the simplest possible core of one proton. The radius of the nucleus is approximately m compared to the hydrogen atom's radius in general, which is approximately m. Rutherford thought that the hydrogen core could be a basic unit for a positive charge. He was the first to use the label proton hydrogen core. The normal position of the atom is neutral. However, atoms can gain or lose electrons, in which case they become electrically charged atoms called ions. Since protons are safely locked away in the nucleus of the atom, only electrons are transferred from one body to another during the electrification of a normal object. Edvantage Interactive 2013 ISBN Chapter 5 Electrostatic PROJECT 263 10 Object electrification Image shows what happens when a vinyl plastic strip is rubbed on wool. Vinyl has a stronger affinity for electrons than wool. When vinyl contacts wool, some electrons leave the wool and go to the surface of the vinyl. It leaves vinyl with excess electrons, so it has a negative charge. Wool, lost electrons, is a positive curse. Image Charging a vinyl rod with wool: vinyl becomes negatively charged and the wool becomes positively charged. Similarly, if acetate plastic by rubbing with cotton, the cotton obtains electrons from acetate. Acetate becomes positively charged, while cotton becomes negatively charged. All experiments show that there is no electrical charge creation or destruction during electrification. All that happens is moving electrons from one body to another. Under the law of protection fees, the electric charge has never been established and has never been destroyed. Electric charging, such as pulse and total energy, is the amount saved. Electrostatic or triboelectric series Regardless of whether an object loses or acquires electrons when rubbed by another object, depends on how tightly the object holds on the electrons. The electrostatic or triboelectric series lists the various objects depending on how tightly they c their e-wires are (Figure 5.1.4). The higher the site on the list is, the stronger it is to hold on to the electrons. The lower the list of objects is, the weaker it is to hold because of the electrons. This means if we rub wool and amber together, the electrons will be moved from wool to amber. As a result, wool is positively charged and amber is negatively charged. Hold electrons tightly vinyl plastic wrap amber cotton paper silk fur wool glass hands + Hold electrons loosely Image Electrostatic or triboelectric series 264 Chapter 5 Electrostatic PROJECT Edvantage Interactive 2013 ISBN 11 Wires and Insulator wires are materials that allow charged particles to pass through them easily. Metals such as silver, copper, and aluminum are excellent conductors of electricity, but all metals perform to some extent. Atoms of metals are one or more external electrons that are very loosely connected to their nuclei so loosely attached that they are called free electrons. In Figure 5.1.5, the metal bar is supported by a plastic bucket. Plastic does not drive electricity. A negatively charged strip of vinyl is allowed to touch one end of a metal rod. When vinyl touches metal, some excess electrons are taken to the rod, so it becomes negatively charged as well. A negatively charged strip pushes away excess electrons to the far end of the metal rod. Initially neutral metal sphere, which hangs from the silk string, attracts a charged rod. When the sphere touches a negatively charged rod, some of the excess electrons are taken to the sphere. Since the sphere now has the same charge as the rod, it is pushed from the rod. Figure Electrons move from vinyl strip to metal rod and to ball. Now both the rod and the sphere are too electrons. If the vinyl strip is taken away, the rod and sphere will retain its negative charge and the sphere will remain in its repelled position. On a dry day, it can stay there for many hours. If the metal bar is replaced by a glass or plastic rod of similar dimensions, the metal ball does not move. This is because glass and plastic are insulators. Isolators are materials that the flow of particles through them. Examples of good insulators are plastic, amber, porcelain, various textiles, sica, sulfur and asbestos. Carbon form diamond is an excellent but very expensive insulator. Carbon in the form of graphite is a good conductor. Non-metals, such as silicon and selenium, find many applications in transistors and computer chips due to their semiconductor behaviour. It is easy to place a static charge on the insulator, because electrons are transferred only when two objects come into contact. When overcharges go up at some point on the insulator, the fee won't flow away so it stays static. Edvantage Interactive 2013 ISBN Chapter 5 Electrostatic PROJECT 265 12 Charging with a guide Electroscope is a device designed to detect excessive electrical charges. In Figure 5.1.6, the strip of positively charged acetate is sufficiently close to touch the electrical set neutral with metal-salted. When they touch, free electrons on the surface of the conducting sphere will be tied to positively charged acetate plastic. Acetate will get some electrons, but its total fee will remain mostly positive. However, the sphere now has a positive fee, so it is repushed by a strip of acetate. We say that the sphere is accused of contact or directing. You could just as easily charge the sphere negatively by touching it with a charged vinyl strip. Illustration Charging with induction objects can be charged without touching at all, in which case we call it induction. There are many ways to do this. The picture below shows one way. Two metal beads are on insulated stands and touch each other. A positively charged acetate strip is brought off two spheres, but it does not touch them. Free electrons from the right sphere are tied to the left sphere with a positive acetate strip. Now the right bullet is ejected away using an isolated support stand. Tests with an electroscope will show that the right ball is charged positively for induction. The left sphere is charged negatively by induction. Induction charge image note that no fee has been created during this procedure. All that has happened is that some electrons have been moved from the right sphere to the left sphere. The total fee is still the same as the charge attempted before the induction. The net fee is still zero. 266 Chapter 5 Electrostatic project Edvantage Interactive 2013 ISBN 13 Investigation using the purpose of conduction and induction Experiment with two different ways to shine on an object part 1 Handling when you charge an object by touching it with another charged object, electrons are carried directly to it. In the process, you are charging with wires procedural rules. Install two aluminum pop cans on or Styrofoam cups, as shown in the figure Styrofoam is an excellent insulator, so it any static charges you place on the cans from escaping to the bench. 2. Place a negative charge on one of the cans as follows: (a) Rub the vinyl strip with wool or fur. You can hear the crackling sound when the vinyl is being charged. Vinyl will have a negative fee for it. (b) Rub the charged vinyl strips over one of the isolated pop bottles. Excess electrons from vinyl will flow on the can, giving can a negative charge. (c) Repeat the process several times to make sure there are many excess negative fees on the can. Image Styrofoam acts as an insulator. 3. Place a positive charge on the other hand, you can: (a) Rub the acetate strip with cotton or paper. This will make acetate positively charged, as electrons flow from acetate to cotton. (b) Rub the acetate strip on the second foot. A positively charged acetate strip will attract electrons from the second metal pop can, making it can be positively charged. (c) Repeat this process several times to make sure that the second can have a lot of positive fees. 4. Do not touch the metal seducaries. Touching only their insulated styrofoam base, move the cans against each other until they are about 3 cm apart. 5. Lower the graphite or pith ball between two opposing charged ladders. Write down what you see happening. 1. Question referred for a preliminary ruling What fee was (a) the first one can start? b) the second can start? (c) graphite ball before it was lowered between legs? 2. Explain what happened to the graphite or pith ball during the experiment. Describe what happened to the electrons going to and from the three objects involved. 3. Why does the action eventually stop? Edvantage Interactive 2013 ISBN Chapter 5 Electrostatic PROJECT 267 Part 14 Part 2 Induction Charging Imagine that you only have a negatively charged strip, but you want to place a positive charge on another object. If you touch another object with a negatively charged strip, you charge it for negative driving. However, if you use the induction method, you can give it a fee, which is the opposite of charging for the charging authority. procedural rules. Place a pop can on or styrofoam cup. 2. Charge the vinyl strip negatively. 3. Bring charged vinyl strips around and parallel pop can, but don't let the vinyl strip touch the can. 4. Briefly touch the cane with your finger, and then remove it and the vinyl strip completely. What do you think the fee is for? Repeat steps 2 to 4 until the same result can be obtained three times in a row. 5. Arrange the procedure to check for yourself whether the fee per foot is positive, negative or neutral. 1. Question referred for a preliminary ruling Before you set your finger off the can, (a) what charge was on the vinyl strip? b) What fee was the side can at the vinyl strip? c) what fee was on the other side can? 2. Your finger may carry electrons on or from your body. In this experiment, were electrons taken to the can of the body or from can your body? 3. (a) What was the final fee per can? b) Was this fee made from a strip of vinyl? c) How can I get this fee? 268 Chapter 5 Electrostatic PROJECT Edvantage Interactive 2013 ISBN 15 5.1 Review Questions 1. What are the similarities and differences between the properties of electron and proton? 4. Draw a series of diagrams to show how an object can take a positive charge using only a negatively charged vinyl strip. 2. Describe the difference between a positive charge and a negative charge in terms of electrons. 3. Draw a diagram to show how an object can take a negative charge using only a negatively charged vinyl strip. 5. Why do clothes sometimes be static on them as soon as they come out of the clothes dryer? Edvantage Interactive 2013 ISBN Chapter 5 Electrostatic project 269 16. What will be the cost of a silk scarf if it is rubbed with glass? With plastic wrap? 8. What would happen if the vinyl strip in the picture was replaced by a positively charged acetate strip? Why? 7. A charged rod is brought to a pile of tiny plastic areas. The spheres are attracted by a charged rod and then fly off the rod. Why is this happening? 9. Outline the method by which you can determine for sure whether to charge for your comb after comb your hair is positive or negative. 270 Chapter 5 Electrostatic PROJECT Edvantage Interactive 2013 ISBN 17 5.2 Electric Force Warm Up You have to pay acetate strips and some confetti. How could you use these two pieces of equipment to prove which force is a stronger gravitational force or electric force? Charles Coulomb When you observe two objects that are tied or pushed due to electrostatic charge, you notice the non-comming forces in action. Both objects are exposed to each other without touching. The force of one insentor body can be measured. The force was originally determined by the French scientist Charles Coomb (. He used an apparatus similar to the Henry Cavendish gravitational force apparatus to work the relationship between these variables: strength, distance and amount of charge. The following illustration shows a setting similar to that used by Coulomb. Image Coulomb uses apparatus like this to study the relationship between variables in strength, distance, and quantity charge. In Coulomb's apparatus, the torque caused by the rejet of two similarly charged spheres caused the length of the vertical wire to twist over the angle. The twist amount was used to calculate the force of deterrence between the two charged areas. The apparatus is called the horse's balance. Edvantage Interactive 2013 ISBN Chapter 5 Electrostatic PROJECT 271 18 Coulomb's Law Coulomb could not directly assess the cost of the spheres. However, he

found you can change the relative amount of the charge as follows: One sphere has an unknown charge Q for it, and the other identical sphere is zero charge. If you touch both of them together, both domains will charge $1 Q$. It is assumed that the 2 excessive fees in the original sphere will be divided equally by the second, identical sphere. This additional fee distribution can be repeated several times to get spheres with a charge of $1.4 Q$, $1 Q$ and so on. The experiments between Coulomb and Others led to the conclusion that the force of attraction or abstinence between the two-point charges depends directly on the excess fee product on the authorities and inversely from the distance square between the two points. It is called the Coulomb Law and is written symbolically as follows: $Q_1 Q_2 F = k \cdot r$ The volume of the proportionality constant k depends on the units used to measure the excess fee. If the unit of measure is a simple charge (as per electron or one proton), q would be measured as simple charges, and $Q Q$ constant $k = k$ is the value $r = N m / 2$ (elem. charge) 2 If the unit of measurement for excess charge is coulomb (named charles Coulomb), then Q will be measured coulombs (C) and constant $k = N m^2 / 2 C^2$ Value k was developed after Coulomb's time. At the time he did his experiments, there were still no units for the quantity of fees. When scientists decided on an appropriate measuring unit charge, they named it after Coulomb. Coulomb's wording of the legislation mentions indications of accusations. The cob's legislation applies to very small fee instabilities. If the charging authorities are large in relation to the distance between them, it is difficult to know what value r use. If the bodies have the same spheres, over which the fee is evenly distributed, then you can use the distance between their centers. If the two large, which control spheres, approach each other, the forces between the projectiles will cause excessive charging to be rearranged in spheres so that the toll centers cannot coincide with the mass centers. Coulomb and elementary fee It is now known that coulomb fees are equal to the amount of charge for electrons (if the fee is negative) or to the same number of protons (if the fee is positive). Fee per electron or one proton, called elementary charge, is 1 elementary charge ($e = V = C = 272$ Chapter 5 Electrostatic project Edvantage Interactive 2013 ISBN 19 Sample Problem Pendant's Law What force would be done with a 1.00 C positive charge per 1.00 C negative charge, which is 1.00 m away? What to think about 1. The two charges are separated by distance. This is Coulomb's right issue. 2. Find each charge and distance and remember to follow the sign. 3. Resolve. 4. As you can see, C charges attract 1.00 C charge 1 m away with a force of nearly 10 billion newtons! Coulomb actually has a very high fee. How do $F = k Q_1 Q_2 / r^2$ ($N m^2 / C^2$) ($1.00 C$) ($1.00 C$) ($1.00 m$) ($2 F = N = P$ Practice Problems Coulomb's Law 1. A small metal ball with μc (micro-colours) are annexed to other metal ball charges at 2.10 μc . The distance between the two spheres shall be 3.7 cm. Find the amount of one charge force that acts on the other. Is it the force of attraction or deterrence? 2. What is the distance between two C and C charges experiencing the N force? 3. The push-back force between two identically charged small spheres shall be 4.00 N if they are 0.25 m apart. What is the charge in each area? Give your answer to microcoulombs (μc). ($1 \mu c = 10^{-6} C$) Use $k = N m^2 / C^2$ 2. Edvantage Interactive 2013 ISBN Chapter 5 Electrostatic PROJECT 273 20 Electrical Forces Due to multiple point charges If more than two charges are in the same area, the force of any of the charges can be calculated with the vector adding forces to it with one of the others. Remember that electrical power is a vector and adding vectors means both size and direction. Sample Problem 5.2.2(a) Three Collinear Charges Three tiny spheres are inged in a row, as shown in figure 1 and the third sphere is 4.00 cm apart and is charge $Q = C$ and $Q = C$. A negatively charged sphere is placed between two positive charges. The cost for this sphere is $Q = B = C$. What is the net force on the negative sphere? Number Three collinear charge What to think about 1. The cost of sphere B is negative and the cost of the A-ball is positive. This means that the strength between the two fees is attractive. The same force between ball B and C. For this problem, the right will be positive. 2. Determine the net force in area b by adding two force vectors. How to do number $F_{net} = F_C + F_A + F_B$ To B C Q B A $F = k + k + 2 r r Nm / C (C) F = (0,20 m) (0,20 m) Nm / C (C) + (0,20 m) (0,20 m) 3$. The net force in area b shall be 0.25 N to the right. $F = N + (0,99 N) = 0,25 N$ 274 Chapter 5 Electrostatic project Edvantage Interactive 2013 ISBN 21 Sample problem 5.2.2(b) Three payments in a triangle Three small spheres with identical charges $+5 \mu c$ are located at the corners of an equilateral triangle with edges 0.20 m long. What is the net force in any of the charged spheres? What to think about 1. The net force per payment will be the sum of the two repellent force vectors generated by two other identical charges. How to do this Let the three charges are A, B, and C. Figure shows their location on the triangle, and vectors representing the forces acting on A with projectiles B and C. 2. First, calculate the force generated by charging charged c Note A: The k value of the kolumb constant for this problem is rounded to Nm^2 / C^2 2.3. F C size on A and F B to A are the same, but their directions are not. Both forces are vectors, and they result in a vector and the result is by means of vector addition as shown in Figure F = $F_{net} = F_C + F_B + F_A$ to A figure $F = k C^2 / Nm^2 / C^2 (C) F = (0,20 m) (0,20 m) F = 5,63 N$ Net F_{net} FC to A sin $120 = 30 F$ C to A = sin $120 = 0,866 (5,63 N) = 9,80 N$ gr 30 (0,500) The direction of the grid force is on the line that crosses angle A, as shown in the figure you can solve the net force in several ways. You can use a scale diagram. Pythagorean theorem cannot be used precisely because the vector triangle is not a right-angled triangle. You could divide it into two right-angle triangles, drawing a line that bisecting 120 angles. The simplest solution is to use the sine law on the power triangle image Edvantage Interactive 2013 ISBN Chapter 5 Electrostatika PROJECT 275 22 Practice Problem Three charge Triangle 1. Small metal ball A with negative C charge is 3.00 cm to the right of another similar Sphere B with a positive C charge. The third sphere with a positive C charge shall be 1.50 cm directly above the second charge, as shown in figure (a) Calculate the net force of the (a) ball. (b) Calculate net force b for area B Calculate net force c 276 in field Electrostatic substance PROJECT Edvantage Interactive 2013 ISBN 23 Investigation Coulomb's Law Purpose To investigate how the force between two electrically charged bullets varies depending on the distance between two spheres Introduction Direct force measurement generated by two charged spheres to each other is complex, but an indirect method can be used to compare forces at different distances. Figure 5.2.6 (a) shows a small graphite-coated ball mounted on an insulating stand. This sphere is charged with touching it with a charged strip of acetate. The acetate strip also collects a movable suspended ball. For the charges imposed on bullets (Q A and Q B), the duties imposed on the whole experiment should remain unchanged. Figure 5.2.6(a) is the distance between the ball centres and d is the displacement of the moving ball when they are pushed back by a similarly charged fixed ball. Vectors, representing gravitational forces, electrical forces and tension in a series, form a right-angled triangle. It should be noted that this force triangle is similar to the displacement triangle in Figure 5.2.6(a). As the triangles are similar, $F_E mg = d L$ Therefore, $mg L D E = Since m, g and L are constant during the experiment, we can write that $F_E = constant d$ or $F_E \propto d$ Since F_E is proportional d , we can use d as a measure of electrical force between two torched bullets at different distances r . Edvantage Interactive 2013 ISBN Chapter 5 Electrostatics PROJECT Edvantage Interactive 2013 ISBN 25 5.2 Overview Questions When calculating your answers use the following figures: $k = Nm^2 / C^2$ 1 elementary charge = $C = 1 C$ = elementary costs 1. What will happen to the force between two charges Q 1 and Q 2 separated by a distance r if: a) one of the charges has doubled? (b) the two charges are doubled? 3. What is the force of the push-back between two structures carrying 6.0 μc charge and separated by 1.0 μm ? 4. What is the power of the binding between proton and electron hydrogen atoms when they are in apart? (c) the distance has doubled? (d) the distance has tripled? 5. The mass of one electron shall be kg. How many coulombs charge would be there for 1 kg of electrons? How much force does this charge act on another 1 kg of electrons, which are 0.6 miles away? (This is a strictly imaginary situation!) e) both charges have doubled and the separation distance has doubled? (f) the two charges are doubled and the separation distance is halved? 2. What force should a positive charge be applied to a 1.00 μc positive projectile at a distance of 1.00 μm ? 6. Two small beads are located at a distance of 0.50 m from each other. Both have the same fee for them. If the scary force is 5.0 N, what is on the ball in the chest, μc ? Edvantage Interactive 2013 ISBN Chapter 5 Electrostatic PROJECT 279 26 7. Three charged objects are located in the corners of an e-edge triangle with edges 1.0 m long. Two of the objects are 5.0 μc each. The third object gives a fee of 5.0 μc . How the resulting force acting on an object of 5.0 μc ? Let's say all three objects are very small. 9. Discuss whether you think gravity could play a major role in holding atoms together. See the results of question 8. Calculate the gravitational force between the proton and electron m interval. Compare this force with the electrical force calculated in question Imagine you could insert 1 g of electrons at a distance of 1.0 m from another 1 g of electrons. (a) Calculate the net force of the (a) ball. (b) Calculate net force b for area B Calculate net force c 276 in field Electrostatic substance PROJECT Edvantage Interactive 2013 ISBN 23 Investigation Coulomb's Law Purpose To investigate how the force between two electrically charged bullets varies depending on the distance between two spheres Introduction Direct force measurement generated by two charged spheres to each other is complex, but an indirect method can be used to compare forces at different distances. 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but since then, therefore, since the electric field $F = qE$ and $F = qV/d$ and its $Q = CV$ and $V = Ed$ The specified electron beam accelerates with a certain accelerated voltage, $1, Q, m, 2$ and t will remain unchanged and it can be assumed that $y = \text{constant}$. E , or that $y = E$. Also, whereas $E = V\delta/d$ where d is the distance between the deflection plates and $V\delta$ is the deflection voltage, so the deviation is proportional to the deviation voltage $V\delta$. Note: the deviation you measure on the screen is not y . When the beam leaves the deflection plates, the electrons move in a straight line to the fluorescent screen. From the fact that you see deviations δ . Fortunately, it can be demonstrated that the screen deviation δ is directly proportional to the true deformation ranges in the plates.* Since δ is proportional to y and y is proportional to the $V\delta$, so $\delta \propto V\delta$. This means that the light deviation measured on the screen is the voltage measure applied to the deflection plates. The toy ray tube (CRT) can therefore be used as a voltmeter. Edvantage Interactive 2013 ISBN Chapter 5 Electrostatic PROJECT 309 56 *Shows that δ is proportional to V . Consider electrons flowing into deflecting plates at horizontal speeds v (Figure 5.5.7). The electron moves the distance $x = l$, at time t . Time is related to distance and speed as follows: $t = x/v$, so the time spent by the electron between the two plates is l/v . Figure Electron beam deviation between the two plates. Now take into account the vertical deviation of the electron when passing through the deforming plates. Since the electric field and therefore the force on the electron are the same, the acceleration is the same, and $y = 1/2at^2 = 1/2a[x^2/v^2t^2]$. This is the equation of parabola. To find the slope of the parabola at the moment when the electron reaches the A (where the deflecting plate ends), we need to use calculus: $dy/dx = a \cdot x = a(1) = 2v/v \cdot x$. When the electron leaves A, its slope remains the same as the A. This means that the trajectory of the straight line of the electron can be extrapolated from the A plate from the plate from the plate. It creates two similar triangles, with a height of y and y . The deviations you actually see on the screen are y , and $y = \delta$. In Figure 5.5.7, the deviation in the plates is y . Similar triangles tell us that $\delta = y/l$. Chapter 310 5 Electrostatic projects Project Edvantage Interactive 2013 ISBN 57 Quick Check 1. Crt, voltage V accelerates electrons from rest to high speed. Assuming that all the work done by the electric field is used to give the electron kinetic energy, $1/2mv^2$, what speed do electrons achieve? (The electron box is C and one electron mass is kg.) What part of the speed of light ($c = m/s$) is it? 2. If accelerating voltage V was used to accelerate protons, what speed do they achieve? ($m_{\text{proton}} = \text{kg}$) 3. The CRT shall be used at an acceleration voltage of 750 V to accelerate electrons before passing through the defective plates to which a deviation voltage of 50,0 V is applied. (a) What speed do electrons achieve? b) When electrons move through deflection plates separated at a distance of 2,0 cm, what is the resistance of the electric field between the plates? c) What is the force that will direct electrons when they cross the plates? d) To what extent will electrons accelerate when they cross (e) The length of the plates shall be 5,0 cm. How long will the electrons be between the plates? f) What is the deviation in the direction of electrons when they pass through the plates? g) If the screen is 0,20 m behind the end of the deformation plates, what deviation (δ) you will see on the screen? Edvantage Interactive 2013 ISBN Chapter 5 Electrostatic PROJECT 311 58 Investigation Deflecting beam electrons Using Electric Field Part 1 Target How does deformation of beam electrons at a given speed depend on deflecting voltages applied to deflecting plates between which beam passes? procedural rules. The image is a schematic diagram of a CRT device similar to the one you will use in this operation. On the X-plates will δ variable deviation voltage V voltage. Add X 1, Y 1, and Y 2 with total binding position A 2. The following apparatus shall be used throughout part 1 with an δ acceleration voltage V a 500 V. b) Connect the second wire from the resistor string +50 V binding point to the other end of the resistor string. c) Connect the cord from X 1 to the total base of the resistors (0 V). Lead from X 2 can now be connected to any point string resistors, so you can get a voltage of 0 V, 10 V, 20 V, 30 V, 40 V, (d) Use a voltmeter to check the exact values obtained at each of the terminals (resistor junctions) and to confirm that this arrangement gives a range of voltages that are from each other. Figure 5 Electrostatic projects Project Edvantage Interactive 2013 ISBN 59 3. Put a piece of masking tape on the screen crt, so you can note the state of the beam when different deflecting voltages are applied. Mark the position of the beam for each of the voltages used (0 V, 10 V, 20 V, 30 V, 40 V and 50 V, if the power supply voltage is in place). 4. The values δ V and values shall be δ V the δ the data table. 5. Prepare δ V δ . Conclude question 1. How does δ of deflection voltage $V\delta$? Type an equation for your schedule, including inclination in the appropriate units. 2. How does the deviation depend on the strength of the electric field between the plates? How does your answer stem from the results of this experiment? (Remember how E and $V\delta$ are related.) Part 2 Objective How does the electron beam deviation depend on the acceleration voltage when a constant deflection voltage is used? Instructor Note: See your CRT manual if these instructions on changing V don't apply to the model your students have. procedural rules. Find the zero position of the light, as you did in part 1, and mark it on a fresh masking tape attached to the screen. In Part 1: 500 V and divorced from the defective voltage. This time, leave the deflection voltage at a value of 50 V throughout Part 2. (You'll use zero deflecting voltage to find zero position, of course.) 2. Mark the position of the electron beam on the screen at an acceleration voltage of 500 V. Turn off the power supply. Reset the CRT on again, redirect and measure the new deformation if necessary, when $V = 750$ V. 3. Turn off the power supply. Change V a to 1000 V. Turn on the CRT, reorient and measure δ . 4. Before analyzing your results, use a voltmeter to check the actual voltage at the three A 2 terminals. Measure the voltage between the filament and each of the A 2 terminals. These are more reliable values to use in V than what you read on the apparatus. The labeled values are only an approximate voltage. 5. Tabulating results, then plot a graph δ (y-axis) vs. accelerating voltage V a (x axis). 6. Make an educated guess at the power of the law relationship that exists between δ and V , and a piece of graphics that will tell you what the value n is $\delta = kV^n$. 7. To obtain the expression on the deviation plates v a. Keep in mind that y is proportional to δ . Start by the fact that the time spent in the plates will be F . Electron acceleration is $= Eq/m = m \cdot mv/x$. Also the acceleration required $v = q \cdot v \cdot x$. According to your derivation, how y (and δ) differ with V a? Edvantage Interactive 2013 ISBN Chapter 5 Electrostatic project 313 60 Final question 1 final questions. Write a simple equation relating to deformation (δ) and acceleration voltage (V a). Include a numeric value for the constant of proportionality (slope). 2. If V a were doubled and $V\delta$ been kept the same, what would happen to δ ? 3. If $V\delta$ been doubled and V a remained unchanged, what would happen to δ ? 4. If both V a and V were δ , what would happen to the δ ? Challenge 1. (a) What happens if you apply a low (6,3 V) AC voltage to the x-plates in your CRT? (b) What if you apply ac voltage to Y plates and X-plates at the same time? (c) Read up on lissajous curves electronics book. Use a full-size oscilloscope to display them. 314 Chapter 5 Electrostatic PROJECT Edvantage Interactive 2013 ISBN 61 5.5 Review Questions 1. Two parallel plates are probably 120 V. The division between the two plates is 10 cm. Calculate the electric field between them. 4. Nichrome wire 30,0 cm long is connected by terminals 1,5 V in a dry cell. What is the volume and direction (relative to the terminals) of the electric field inside the wiring? 2. The two electrodes on the spark plugs have an electric field between them at 500 V/m and are separated by m. What is the voltage between the two plates? 5. Two parallel field plates in the ostilostoge 120 V. They are separated by a distance of 2,4 mm. (a) What is the strength of the electric field between the plates? 3. What is the distance between two plates with a voltage of 800 V and an electric field N/C? b) To what extent will the electron (mass in kg) accelerate when it enters the space between the plates? (The electron initially moves in the direction perpendicular to the field lines.) Edvantage Interactive 2013 ISBN Chapter 5 Electrostatic project 315 62 6. Between two charged plates, balancing the gravity force downwards with the same size of electrical force upwards, holds a small mass of plastic spheres in kg. a) What is the amount of electrical force acting on the sphere? 7. Two parallel plates have a potential difference between 2000 V and are m at a distance from each other. The proton is released from the positive plate at the same time as the electron is rejected from the negative plate. Compare and describe their speed and kinetic energy as they strike the opposite plate. b) If the application voltage on the plates is 30,0 V and the plates are separated by a distance of 1,47 mm, what is the sum of the overcharge on the ball? 8. Subatomic, charged C particle is accelerated from rest through V voltage. If the final particle speed is m/s, what is the mass of the particles? c) If the upper plate is positively charged, is the bead charge positive or negative? 9. The resting position of the bracket shall be accelerated between two parallel plates with a possible difference of 600 V, as shown below. What is the maximum speed of protons? d) What kind of elementary charged particles are on the ball, and how many of these extra elementary particles are on the ball? 316 Chapter 5 Electrostatic projects Project Edvantage Interactive 2013 ISBN 63 10. In the diagram below, the charged Sphere C and kg mass are held in the middle of two charged plates and balanced by gravity and electrical forces. What is the voltage V suitable for these plates? 11. CRT, if the deviation on the screen is 2,4 cm, if the acceleration voltage is 480 V and the deflection voltage is 36 V. What deviation (δ) do you see on the screen if the acceleration voltage is 960 V and the deflecting voltage is 18 V? Edvantage Interactive 2013 ISBN Chapter 5 Electrostatic project 317 64 Chapter 5 Review of questions 1. Two equal charges Q shall be separated by a distance r . The aidy force between them is F . (a) What will be the force if both charges are doubled? 3. (a) A tiny plastic ball is in the middle between two metal plates at a distance of m. When the battery is connected to the plates, the ball experiences electrical force N . How much work is needed to move the bear ball from one plate to another? b) What will be the force if the fee remains in Q , but the distance between them is reduced to 1,2 r? b) If the ball charge is 0,20 μ C, what is the battery voltage? c) What strength will it be charges are increased to $4Q$, and the distance between them is reduced to 1,4 r? 4. What is the scary force between two alpha particles that are 1,0 mm apart? (The alpha particle is a helium ion, He^{2+} , so it carries more than two elementary positive charges.) 5. How far apart are the two protons if they repel each other by a force of 1,0 mn? 2. A negatively charged rod is brought to a suspended metal ball. The sphere is grounded by touching it with your finger. The finger and rod have now been removed. What will be the cost for the sphere, positive or negative? Describe what happened. 6. If the body with charge C experience a force of 1,0 N electric field, what is the strength of the electric field? 318 Chapter 5 Electrostatic projects PROJECT Edvantage Interactive 2013 ISBN 65 7. Proton is placed in the N/C electric intensity field. (Mass of protons = kg; proton projectile = C) 10. The defective voltage applied to crt plates is 50,0 V. The plates are 1,2 cm apart. (a) What is the strength of the electric field between the plates? b) What force will the field have on the electron passing between the plates? 8. What is the electrical field strength and direction in the middle between the 75 μ C charge and the 25 μ C charge if the charge is 2,0 m from each other? (c) At what speed and direction will the electron accelerate? (Electron mass = kg) 11. An acceleration voltage of 750 V produces screen deviations of 4,2 cm per CRT. If the deflecting voltage is maintained unchanged, but the acceleration voltage is increased to 1000 V, what will the deviation become? 9. The strength of the electric field between two plates, which is 3,0 cm apart, is N/C. What is the voltage between the plates? 12. If the acceleration voltage CRT is 500 V and the deflection voltage is 15 V, the beam deviation on the screen shall be 1,2 cm. If the acceleration voltage is changed to 750 V and the deflecting voltage is changed to 45 V, what will the deviation be? Edvantage Interactive 2013 ISBN Chapter 5 Electrostatic project 319 66 13. How much work should be done to move a positive +2,0 μ C projectile from infinity to a point 1,2 m from a positive point q ? P is 3,0 cm from both charged solders as shown below. 14. What is the electrical potential for infinity m away from proton? 15. Two-point μ C projectiles shall be separated at a distance of 1,0 m. If the force between them is F , what will be the force if 4,0 μ C is removed from one point charge and transferred to the other point of charge? 17. Protons are accelerated by a possible difference of 6,0 MV (megavolti) and then carry head-on collisions with atomic nuclei with a charge of +82 elementary charges. What is the closest distance approach between protons and nuclei? Let's say the kernels are stationary. 18. How much must do to make three protons from infinity to distance m from each other? 320 Chapter 5 Electrostatic project Edvantage Interactive 2013 ISBN 67 Chapter 5 Additional practice Answer questions 1 and 2 using the scheme below. 1. What is the direction of electric field B, located between a couple of opposite charged plates? 2. Where is the electrical field strength between the plates strongest and weakest? 3. What is the direction of electric field P due to the point charge in the Q 1 and Q 2 chart below? 7. An atom that carries an excessive C charge accelerates from rest with a possible difference of 750 V. It reaches a maximum speed of m/s. What is the mass of an atom? 8. What increases the power potential of energy when an alpha particle with C charge is imported from infinity to c fixed charging distance? 9. Calculate the electrostatic tightening force between a positive C charge and a negative C charge if they are at a distance of 0,30 m from each other. 10. When a charged object is accelerated with a possible difference of 500 V, its kinetic energy increases from J to J. What is the amount of the object charge? 11. How fast will the electron move if it is accelerated from resting, vacuum, with a possible difference of 200 V? 4. Which of these units is the measure of electric field strength? (a) N/A b) J/C c) N/A m (d) N/kg e) V/m 5. The electric field strength 1,0 m from the point of charge is N/C. What will the electrical field strength be at a distance of 2,0 m from the same point charge? 12. Two parallel plates shall be 4,0 mm apart. If the possible difference between them is 200 V, what is the strength of the electric field between the plates? 13. The electron enters the space between two otherwise charged parallel plates as shown below. What is the volume and direction of electrostatic forces that operate on the electron when it is between two plates? 6. The electron beam in the totod beam shall be accelerated against the anode by accelerating the voltage of 100 V. After passing through the anode, the electrons are diverted as they pass between the two otherwise charged parallel deflecting plates. The deviation observed on the screen is δ . If the acceleration voltage is increased to 400 V, what deviations will be observed on the screen? Edvantage Interactive 2013 ISBN Chapter 5 Electrostatic project 321 68 14. Charging C in the diagram shown in q diagram shall be the opposing forces generated by 5,0 N and 11,0 N Q 1 and Q 2 respectively. What is the volume and direction of electric field strength in place q? 16. What is the electrical potential of an electron in relation to the infinity that is m from the hydrogen atom proton? 17. What is the electrical potential of P due to charges Q 1 and Q 2? 15. How much work should I do to move the charge Q 2 = C from A to B, as shown in the diagram below? The second charged object has a charge of Q 1 = C. Chapter 5 Electrostatic PROJECT Edvantage Interactive 2013 ISBN 69 Chapter 6 Electrical Circuits By the end of this chapter, you should be able to take the following actions: Apply oms law to various co-existing circuits Apply Kirchhoff's laws to various co-start circuits Properly use ammeters and voltmeters Solve a range of problems related to the current, resistance, electrical potential difference, electrical power, and efficiency By the end of this chapter, you should know what these key terms mean: ammeter ampere normal electrical current electrical current electrical power electrical power (EMF) equivalent resistance internal resistance parallel circuit resistance series circuits terminal voltage voltmeter Until the end of this chapter, you should be able to use and know when to use the following formulas : $Q = V = IR$ $P = IV$ $V_{\text{terminal}} = \pm$ There t! These high voltage transmission lines play an important role by transporting electricity to your home and school, where energy is used to power a range of electrical appliances, including hybrid vehicles. Edvantage Interactive 2013 ISBN Chapter 6 Electrical Circuit PROJECT 323 70 6.1 Current Events History Warm-up Your Teacher will give you a light source, energy source and wiring. How many different ways can you arrange these three items so that the light source continues? Draw each chain. Galvani's and Volta experiments Up to 200 years ago, the idea of producing a stable current of electricity and putting it into use was non-existent. The discovery of the way to create the flow of electrical charges was, in fact, accidental. In 1780, at the University of Bologna, Italian professor of anatomy Luigi Galvani () was a dissecting frog. First, he noticed that when a nearby static power generator made a spark while a metal knife was touching frog nerves, the frog's legs would jump because its muscles contracted! Galvani continued to look for the conditions that led to this behavior. During the investigation, Galvani discovered that when two different metal objects (such as a brass hook and iron strut) touched each other while touching frogs exposed to flesh, the same contraction of the frog's legs occurred. Galvani thought that the source of electricity was the frog itself, and he called the phenomenon of animal electricity. Another Italian scientist, physics professor Alessandro Volta () of Pave university, who is going to test galvani animal electricity theory for himself. A long time ago, Volta discovered that the source of electricity was in contact with two different metals. The animal (frog) was accidental. If any two different metals are submerged in a conductive solution of acid, base, or salt, they will create an electric current. Volta was able to prove that some pairs of metals worked better than other couples. Of course, no ammeters or voltmeters were available these days to compare currents and voltages. One way Volta compared was to observe the reaction of muscle tissue in dead frogs. Neither Galvani nor Volta could explain their observations. (There is no truth to rumors that the frog leg jumped since 1780 was a year long.) Many years later, it was learned that the radio waves generated by the ignition generator caused a current of a metal scalpel that was penetrating the frog, even though the scalpel was some distance away from the generator! Volta finally invented the first practical electric battery. Zinc and silver discs, separated by paper pads soaked with salt water, acted as electrical cells. Stake one above the other, these elements became a battery that gave more power than one cell. 324 Chapter 6 Electrical CircuitS PROJECT Edvantage Interactive 2013 ISBN 71 Cells and Batteries There are many types of electrical cells that exist today. Usually, when you buy a battery in a store, you actually buy one cell. Strictly speaking, the battery consists of two or more cells connected together. The nine volt batteries used in portable radios, recorders, calculators and smoke alarms are true batteries. When you open a discarded 9 V battery, you'll see six small 1,5 V elements joined together, one after the other, called a series. The picture below shows one type of volta cell (named Volta). Two rods, called electrodes, are made of carbon and zinc, as in a traditional dry cell, but they are immersed in ammonium chloride solution (NH 4 Cl). In a real dry cell, paste containing NH 4 Cl, sawdust and other ingredients is used. The chemistry of the voltaic cells will be left in your chemistry courses. The reaction that happens, however, has consequences in removing electrons from the carbon electrode (making it positive) and adding them to zinc (making zinc negative). In a real dry cell, the outer body of the cell is made of zinc. Zinc is dissolved away because the cell is used and can eventually leak its contents. Image Voltaic cells There are many types of cells and batteries. Rechargeable batteries can use nickel and cadmium, molybdate and lithium, or lead and lead oxide (as in a standard car battery). There are many types of cells on the market today, but they all produce electric current when connected by a controlled road. Electromotive Force Carbon-Zinc in a dry cell, forces the resulting from the chemical reaction in the cell to drive the charge to the terminals, performing work to overcome the resentful forces. Working on the charge increases their potential energy. The possible energy difference between terminals is 1,5 J for each separated charge. We say that the possible difference is 1,5 J/C, or 1,5 V. For a cell or battery that doesn't play current, the potential difference is its maximum value, called electromotive force (EMF). It is given a symbol ϵ . The dry cell EMF has a 1,5 V. Nickel-cadmium cell is usually marked at 1,2 V or 1,25 V, but barely nickel-cadmium battery will have an even higher EMF than its labelled rating. Lead storage batteries are emfs 2 V each. Six of these elements connected to the battery provide a total EMF 12 V. Edvantage Interactive 2013 ISBN Chapter 6 Electrical Circuit PROJECT 325 72 Electric current When the electrical charge flow, we say that there is a current. The current will persist as long as a continuous driving path is established to flow from the emf source and back to it, as shown in the figure Figure Electric charge flow as a current through the conductor, unless there is a break on the road. Electric current is defined as the amount of electrical charge that passes the point of the circuit in one second. If the amount of charge Q goes point in the circuit at time t , then the Q average current I through this point is $I = Q/t$. The current could be perfectly logically measured in the culmination per second (C/s), but this unit is called ampere (A) by André Ampère (a French physicist). 1 A = 1 C/s One ampere is the type of current that exists in the 100 W bulb lamp in your home. (100 W lamp in 110 V chain retracted approximately 0,9 A.) For electrons, 1 A = electrons/electrons in the current direction, André Ampère arbitrarily defined the direction of the current in order to move positive charges between two points where there are potential energy differences. In many simple chains, we now know it's negative payments (electrons) that actually move. For solid conductors, positive charges are blocked in atomic nuclei that are set in their location crystal. Loosely attached electrons can move through the conductor from the atom to the atom. However, in liquids and gases, the charge flow can consist of positively filled ions, as well as negatively filled ions and electrons. Throughout this book, we will use the usual current direction: the direction in which positive charges could move between two points, where there is a possible difference between the points. Drift Velocity How fast do electrons move the wire in a current like 1 A? When the switch is closed in the circuit, the current effect can be determined immediately throughout the circuit. This can lead you to conclude that electrical charge (usually electrons) travel at very high speeds through the circuit. Actually, it's not! You can calculate the 326 Chapter 6 Electrical Circuit PROJECT Edvantage Interactive 2013 ISBN 73 average drift speed electrons in a given set of conditions can be calculated. Within the length of the metal wire, there are many, many loosely attached electrons (sometimes called free electrons). These electrons move around a lot like molecules in a container of gas could move. Let's look at the movement of electrons in the silver wire. Silver is a great conductor. Let's say that each silver atom has one free electron in a piece of wire. When the wire is connected to the emf source, the possible difference (voltage) will cause the electrons to shift from the negative terminal source EMF to the positive terminal. This movement is superimposed on the accidental movement of the electron, which occurs all the time with or without the source of the EMF. The journey made by electrons from negative to positive terminal is not smooth (as it is, for example, crt vacuum). Electrons in metal wire collide with positive silver ions en route and put some of their kinetic energy into silver ions. The increased heat energy of silver ions will appear as a temperature rise silver conductor. Let's say the silver wire current in the picture is 1,0 A. Then there are electrons passing the observer every second. This is because $1,0 A = 1,0 C/s = 6,24 \times 10^{-19} C/s$. Calculating the average drifting rate of an electron through a silver wire To calculate the average drifting rate of an electron, we start by once again assuming that there is one free electron for each silver atom, a reasonable assumption, because the chemist tells us that silver usually makes up the ions with a charge of +1. All we need to know is how long the silver wire image would be silver atoms (and therefore free electrons). We can find this in three stages, as follows: 1. Silver mass required to have free electrons is mass = atoms $108 \text{ g mole} = \text{atoms mole} \cdot 10^{-3} \text{ g}$. The volume of silver wire required to release the electrons is 3 masses g volume = cm³. 2. Density of silver wire in the figure shall be 1,0 mm or 1,0 $\times 10^{-3}$ cm. Its cross-sectional area is πr^2 , so the length, l , can be found as follows: volume length, $= \text{area} (cm^2 \text{ cm}) = \text{cm}$. Its length cm contains electrons, and this many electrons pass observer 1 s, the average drift speed out electrons is cm/s, or mm/s! This, of course, applies only to the conditions indicated. If the amount of current, the nature of the material conductor, or the dimensions of the conductor changes, then the drift speed will also change. When you switch on the switch to switch the lamp using the battery, changes in the electric field can travel at the speed of light, but the electrons themselves drift so slowly through the wire under the influence of the electric field. Representing electrical circuitry Image shows a simple electrical circuit. There is a power source or battery, a switch or control, and wires to carry out energy. Illustration Circuits in which it is connected to the circuit, the circuit is said to be in series. The following table lists some of the basic components of a circuit and their symbols.

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